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Addition/Correction

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Correction to "Manganese Catalysts with Bulky Bipyridine Ligands for the Electrocatalytic Reduction of Carbon Dioxide: Eliminating Dimerization and Altering Catalysis"

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Page 5466. In Table 1, the turnover frequency (TOF) values for Mn(bpy-^{t}Bu)(CO)$_3$Br (last column) were reported incorrectly from ref 15. The complete corrected table is shown below.

Table 1. Comparison of Peak i_{cat}/i_p and TOF Values for Both [Mn(mesbpy)(CO)$_3$(MeCN)](OTf) (2) and Mn(bpy-^{t}Bu)(CO)$_3$Br in MeCN (1 mM each catalyst)[a]

	[Mn(mesbpy)(CO)$_3$(MeCN)](OTf) (**2**)			Mn(bpy-^{t}Bu)(CO)$_3$Br		
Brønsted acid	[acid][b] (M)	i_{cat}/i_p[c]	TOF (s^{-1})	[acid][d] (M)	i_{cat}/i_p[e]	TOF (s^{-1})
H_2O	3.5	20	700	3.1	25	120
MeOH	3.2	30	2000	5.8	26	130
TFE	1.4	50	5000	1.4	42	340

[a]Solutions are saturated with (approximately 0.19−0.28 M) and under an atmosphere of CO_2 with added weak Brønsted acids. Data are taken from voltammograms at a scan rate of 100 mV/s. [CO_2] is approximately 0.28 M in dry MeCN, 0.26 M in 3.5 M H_2O, 0.27 M in 3.2 M MeOH, and 0.27 M in 1.4 M TFE.[22] [b][Acid] at highest i_{cat}/i_p for **2**. [c]i_{cat}/i_p values are calculated at equal [acid]. [d][Acid] at highest i_{cat}/i_p for Mn(bpy-^{t}Bu)(CO)$_3$Br. [e]Values taken from ref 15 [Smieja, J. M.; Sampson, M. D.; Grice, K. A.; Benson, E. E.; Froehlich, J. D.; Kubiak, C. P. *Inorg. Chem.* **2013**, *52*, 2484−2491].

This correction to Table 1 changes some of the discussion on the comparison between complex **2** and Mn(bpy-^{t}Bu)(CO)$_3$Br. These changes are described below.

Page 5465. The statement, "Catalyst **2** is more active than the most active Mn bpy catalyst previously reported, Mn(bpy-^{t}Bu)(CO)$_3$Br, under MeOH and TFE. However, **2** is slightly less active than Mn(bpy-^{t}Bu)(CO)$_3$Br under H_2O. Particularly, under identical conditions with 1.4 M TFE, catalyst **2** is ∼1.2 times more active than Mn(bpy-^{t}Bu)(CO)$_3$Br." should be changed as follows: "Catalyst **2** is more active than the most active Mn bpy catalyst previously reported, Mn(bpy-^{t}Bu)(CO)$_3$Br,[15] under all weak Brønsted acids studied. Under similar concentrations of TFE, catalyst **2** is over 10 times more active than Mn(bpy-^{t}Bu)(CO)$_3$Br."

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